Efficient stereoselective synthesis of 1-thio-β-mannopyranosides

Henry N. Yu, Chang-Chun Ling and David R. Bundle*

Department of Chemistry, University of Alberta, Edmonton, Alberta T6G 2G2, Canada. E-mail: Dave.Bundle@ualberta.ca

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An efficient method for the synthesis of 1-thio- β -mannopyranosides is reported. This method employs the simple, easy-to-make 2,3,4,6-tetra-*O*-acetyl-1-*S*-acetyl-1-thio- β -D-mannopyranose as starting material to conduct an *in situ* selective de-*S*-acetylation, and subsequent S_N^2 reaction with an acceptor bearing a leaving group. The high nucleophilicity and slow anomerization of the intermediate thiol allows the synthesis of 1-thio- β -mannopyranosides in a simple and practical manner.

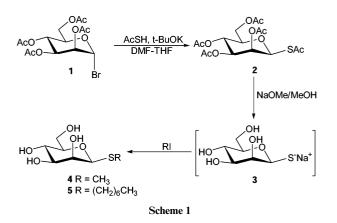
Introduction

Compared with the corresponding O-glycosides, thioglycosides are resistant toward enzymatic hydrolysis and acid degradation,1-4 properties that make them promising candidates as enzyme inhibitors, carbohydrate mimics and potential therapeutics. The chemical synthesis of a variety of 1-thiooligosaccharides has appeared in the literature,^{1,2,5} and recently, increased effort has been focused on the synthesis of thiosialic acid-containing oligosaccharides,⁶⁻¹² However, the synthesis of 1-thio-\beta-mannopyranosides has not been fully explored. Although an enormous effort has been applied to the synthesis of O-linked β-mannopyranosides, their synthesis remains one of the most challenging to carbohydrate chemists.¹³ Sulfur isosteres of these glycosides have considerable utility and, due to the superior nucleophilicity of sulfur over oxygen, the synthesis of S-linked β -mannopyranosides is an attractive option. To the best of our knowledge, the examples of stereoselective synthesis of 1-thio-β-mannopyranosides in the literature are limited, and the few examples reported have been limited to simple aglycones such as thiophenyl or its analogs.^{14,15} In 1984, Defaye et al.¹⁶ prepared 2,3,4,6-tetra-O-acetyl-1-S-acetyl-1-thio-β-Dmannopyranose using an S_N^2 approach by reaction of 2,3,4, 6-tetra-O-acetyl-α-D-mannopyranosyl bromide with tetrabutylammonium thioacetate in toluene. More recently, Crich's group¹⁷ reported a glycosylation approach by reaction of a rather elaborate anomeric sulfoxide, namely S-phenyl 2,3-di-Obenzyl-4,6-O-benzylidene-1-deoxy-1-thia-a-D-mannopyranoside S-oxide, with acceptors containing a thiol group, under the activation of triflic anhydride. Typically, this method allowed

activation of triflic anhydride. Typically, this method allowed the stereoselective preparation of two 1-thio- β -mannopyranosides bearing simple alkyl aglycones in 74–77% yield, and several more elaborate 1-thio- β -manno-disaccharides in 61– 69% yield. These developments in the synthesis of thioglycosides prompted us to explore the potential of 2,3,4,6-tetra-*O*acetyl-1-*S*-acetyl-1-thio- β -D-mannopyranose as a convenient starting material in the stereoselective synthesis of 1-thio- β mannosides.

Results and discussion

2,3,4,6-Tetra-*O*-acetyl-1-*S*-acetyl-1-thio- β -D-mannopyranose **2** was prepared by a simple nucleophilic displacement of the anomeric α -bromide of 2,3,4,6-tetra-*O*-acetyl- α -D-mannopyranosyl bromide **1** with thioacetate. In contrast to Defaye's procedure, we found the reaction worked equally well using the cheaper potassium thioacetate. The reaction was carried out in DMF and the potassium thioacetate was prepared *in situ* by mixing thioacetic acid and potassium *tert*-butoxide, leading to exclusive formation of the 1-thio- β -mannopyranoside **2** (Scheme 1). The reaction can be easily scaled up and compound



2 was obtained in 79% isolated yield on a 30 g scale, following chromatography, or in 63% yield by crystallization.

The β -anomeric configuration of **2** was unambiguously established by different NMR experiments. As a general rule, the homonuclear three-bond coupling constants ($J_{1,2}$) in 1-thio- β -mannopyranosides are always smaller than 1.3 Hz and the heteronuclear one-bond coupling constant (J_{CLH1}) at the anomeric centers range from 148 to 160 Hz; in addition, transferred nuclear Overhauser effect spectroscopy (TROESY) experiments should show dipolar coupling between H-1, H-3 and H-5 within the same pyranose ring while the corresponding 1-thio- α -mannopyranosides shows no such interactions. In fact, for compound **2**, a 1D-TROESY experiment showed strong NOEs between H-1, H-3 and H-5, and a 2D heteronuclear multiple quantum-filtered coherence (HMQC) experiment confirmed the one-bond heteronuclear coupling constant, ${}^{1}J_{CH}$ for C-1 to be 155.6 Hz.

It is well known that the some of the fully deprotected glycosyl 1-thiolates are stable, and anomerizations are so slow that in the case of 1-thio- β -D-glycopyranose and 1-thio- β -D-galactopyranose, the sodium salts can even be purchased from commercial sources. The salts can often be used to conduct $S_N 2$ glycosylations and the corresponding thioglycosides can be obtained in pure anomeric form. In this context, we were interested in determining the stability of the fully deprotected β -manno-1-thiolate **3**. This was carried out by periodically

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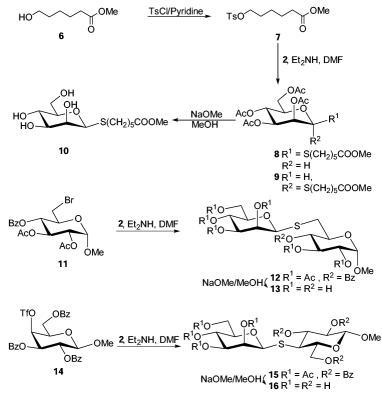
recording the NMR spectrum of a sample of compound 2 in $CD_3ONa-CD_3OD$. The NMR experiment clearly showed that compound 3 was relatively stable, and that anomerization to the α -thiolate proceeded at a very slow pace, such that after three days in this basic solution almost 90% of the compound remained as the β -anomer. Therefore, by exploiting the high nucleophilicity of thiolate 3 addition of simple alkyl halides to the solution should trap the corresponding 1-thio- β -mannosides to give the corresponding glycoside in high yield. In fact, the methyl (4) and heptyl (5) glycosides were both efficiently synthesized in excellent yield ($\longrightarrow 4$, 91% and $\longrightarrow 5$, 97%) by adding iodomethane or iodoheptane directly to a solution of 2 in CH₃ONa-CH₃OH. No 1-thio- α -mannopyranosides were detected.

Glycosides bearing the ω -(methoxycarbonyl)octyl group (often called the 'Lemieux tether') are widely used in the preparation of various carbohydrate conjugates. We were interested in synthesizing a 1-thio- β -mannopyrannoside bearing this tether group. Thus 5-(methoxycarbonyl)pentan-1-ol 6 was treated with toluene-p-sulfonyl chloride in pyridine to afford the corresponding sulfonate ester 7 (75%) (Scheme 2). In their thiosialoside synthesis, Itzstein and co-workers9 reported a convenient procedure to selectively deprotect the S-acetate in the presence of O-acetates via in situ reaction with diethylamine in DMF; the intermediate thiol can then be used directly for attacking a sugar triflate to afford the corresponding thioglycoside. We applied this procedure to prepare compound 8. At room temperature the reaction proceeded rapidly; however, a $1:1 \alpha/\beta$ mixture of glycoside was obtained. When the reaction was carried out at lower temperatures, a longer reaction time was needed to reach completion; however, the ratio of the β -mannopyranoside improved. The optimum result was obtained when the reaction was carried out at -55 °C, and under these conditions the reaction was complete after 48 h and compound **8** was obtained in 64% yield along with its α -isomer **9** in 21% yield. The relatively low β/α selectively may be due to the leaving-group properties of the tosyloxy group in compound 7. The β -mannoside 9 was smoothly deprotected by Zemplen methanolysis (\longrightarrow 10, 91% yield).

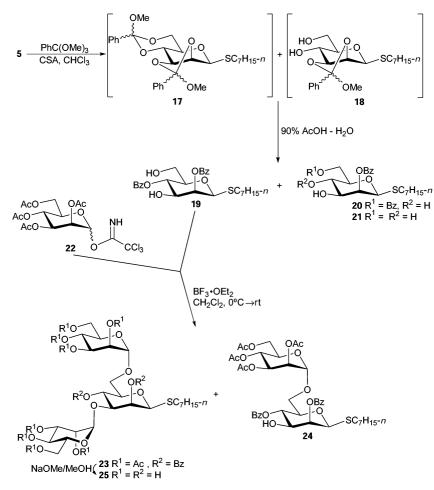
Encouraged by the initial success in synthesizing 1-thio- β mannopyranosides bearing simple aglycones, we proceeded to explore the generality for the preparation of more complicated oligosaccharides. The preparation of a 1,6-linked disaccharide was the first choice. The 6-bromo glucoside (easily accessible) **11** was chosen. When the reaction between **2** and **11** was carried out at -15 °C under conditions reported by Itzstein,⁹ the expected disaccharide **12** was obtained in 74% yield with excellent β -selectivity ($\alpha : \beta, 1 : 15$ as judged from NMR). However, at higher temperature, poor β -selectivity was observed. Deprotection of **12** was conducted under standard Zemplen transesterification conditions, and the fully deprotected disaccharide **13** was obtained in pure form by reversed-phase chromatography in excellent yield (92%).

This success encouraged us to attempt the synthesis of the 1,4-linked disaccharide **15**. Here we employed a galactosyl triflate **14**¹⁸ as starting material; the reaction with **2** gave **15** in 80% yield. The reaction was carried out at -5 °C and the $\beta : \alpha$ ratio was 95 : 5 as judged by NMR. A similar deprotection step as described above was conducted to afford disaccharide **16** in 93% yield after reversed-phase chromatography on C-18.

With compound 5 in hand, we prepared the biologically relevant structure 25-an analog of the core mannotrioside structure located in the glycan chains of all N-linked glycoproteins. In this structure, a β-mannopyranoside is branched at both the 3- and 6-position by α -mannopyranoside residues. The synthesis of the O-linked analogs was reported by Kaur et al.¹⁹ and recently by Lichtenthaler et al.²⁰ In the former case, the O-linked β-mannopyranoside was synthesized from 3,6-di-O-acetyl-2,4-di-O-benzyl-α-D-mannopyranosyl bromide in 27% yield, whilst in the latter case the authors had to create the *O*-linked β -mannopyranoside in a long oxidation–reduction sequence starting from a glucose derivative. In our hands, since 5 can be easily synthesized in large quantity and in pure form, we used it as starting material. Thus, under camphor-10sulfonic acid (CSA) catalysis, two mole equivalents of trimethyl orthobenzoate reacted with 5 to afford mainly intermediate 17,²¹ plus the mono-orthoester 18 (see Scheme 3). Without purification, the mixture of intermediates 17 and 18 was sub-



Scheme 2



Scheme 3

sequently treated with 90% acetic acid-water in order to open the 2,3- and 4,6-orthoester. The 2,3-orthoester opened in a highly regioselective fashion to give the 2-benzoate almost exclusively, whilst the 4,6-orthoester opened in a less regioselective manner to give both the 4-benzoate and 6-benzoate. Thus hydrolysis of 17 afforded the desired 2,4-dibenzoylated product 19 (32%) and the undesired 2,6-dibenzoylated product 20 (40%) while the hydrolysis of 18 led only to monobenzoylated mannoside 21 (13%). Although not of high yield, this approach to 19 was efficient since it only involved a single purification step, and a readily available starting material (5). Compounds 20 and 21 can be recycled to regenerate the starting material 5. The two α -mannopyranosyl units at the 3- and 6-position were subsequently installed after a glycosylation step with imidate 22²² under catalysis by boron trifluoride-diethyl ether. The desired trimannoside 23 was obtained in 51% yield together with the 1,6linked disaccharide 24 (48%). Trisaccharide 23 was fully deprotected by methanolysis to give the free trimannoside 25 in 91% yield.

Conclusions

In summary, we have developed an efficient route for the preparation of 1-thio- β -mannopyranosides using the easily accessible 2,3,4,6-tetra-*O*-acetyl-1-*S*-acetyl-1-thio- β -D-mannopyranose **2** as starting material. The synthesis can generally be carried out in high yield and high stereoselectivity. Considering the significant difficulties in preparing the *O*-linked β -mannopyranosides, this route offers an alternative to the design and synthesis of carbohydrates analogs containing a β -mannoside linkage. The method is being extended to the synthesis of 1-thiol- β -rhamnopyranosides.

Experimental

Optical rotations were measured with a Perkin-Elmer 241 polarimeter at 22 ± 2 °C. $[a]_{D}$ -Values are given in units of 10^{-1} deg cm² g⁻¹. Analytical TLC was performed on Silica Gel 60- F_{254} (E. Merck, Darmstadt) with detection by quenching of fluorescence and/or by charring with sulfuric acid. All commercial reagents were used as supplied and chromatography solvents were distilled prior to use. Column chromatography was performed on Silica Gel 60 (E. Merck 40-60 µm, Darmstadt). ¹H NMR spectra were recorded at 300 MHz (Varian) or 600 MHz (Varian). The first-order proton chemical shifts $\delta_{\rm H}$ are referenced to either residual CHCl₃ ($\delta_{\rm H}$ 7.24, CDCl₃) or internal acetone ($\delta_{\rm H}$ 2.225, D₂O). HMQC NMR spectra were recorded at 300 MHz (Varian) or 600 MHz (Varian). The ¹³C chemical shifts, $\delta_{\rm C}$, are referenced to internal $CDCl_3$ (δ_C 77.00, $CDCl_3$). Organic solutions were dried prior to concentration under vacuum at <40 °C (bath). Reversed-phase chromatography was performed on a Waters 600 HPLC systems, using a Beckman semi-preparative C-18 column (10×250 mm, 5 μ), and the products were detected with a Waters 2487 UV detector or a Waters 2410 refractive-index monitor. Microanalyses and electrospray mass spectra were performed by the analytical services of this department.

2,3,4,6-Tetra-*O*-acetyl-1-*S*-acetyl-1-thio-β-D-mannopyranose 2

Method A. A solution of thioacetic acid (13.4 mL, 187.6 mmol) in dry DMF (130 mL) was cooled to 0 °C under argon, and potassium *tert*-butoxide (12.6 g, 112.6 mol) was added by portions. After stirring of this mixture for 15 min at room temperature, a dark red homogenous solution was obtained. A solution of 2,3,4,6-tetra-O-acetyl- α -D-mannopyranosyl bromide 1 (38.6 g, 93.8 mmol) in anhydrous THF (50 mL) was

added dropwise over a period of 20 min and the reaction was continued for 3 h at room temperature. The mixture was diluted with EtOAc (2000 mL), the organic phase was successively washed with water (3×600 mL) and brine (1×500 mL), dried over anhydrous Na₂SO₄, and concentrated. Chromatography on silica gel (hexane–EtOAc 4 : 1 v/v) gave compound **2** as a white solid (30.1 g, 79%).

Method B. The reaction was carried out as above starting from bromide 1 (35.0 g, 93.3 mmol), thioacetic acid (12.5 ml, 174.9 mmol) and potassium tert-butoxide (11.8 g, 105.1 mmol). After washing of the organic phase with water and brine, the organic solution was dried with Na₂SO₄, decolorized with charcoal, and filtered through a thin Celite pad. Compound 2 (23.9 g, 63%) was obtained by crystallization from EtOAc and hexane; $[a]_{D}^{22}$ -23.6 (c 3.6, CHCl₃); mp 128 °C (from hexane - EtOAc) (Found: C, 47.1; H, 5.1; S, 8.05. C₁₆H₂₂O₁₀S requires C, 47.3; H, 5.4; S, 7.9%); $\delta_{\rm H}$ (600 MHz; CDCl₃) 5.47 (m, 2H, H-1, H-2), 5.24 (t, 1H, J 10.1, H-4), 5.13 (dd, 1H, J 3.5, 10.3, H-3), 4.24 (dd, 1H, J 5.3, 12.5, H^a-6), 4.10 (dd, 1H, J 2.2, H^b-6), 3.80 (ddd, 1H, H-5), 2.34 (s, 3H, SAc), 2.16 (s, 3H, OAc), 2.06 (s, 3H, OAc), 2.02 (s, 3H, OAc), 1.96 (s, 3H, OAc); $\delta_{\rm C}$ (600 MHz; CDCl₃, from HMQC) 79.3 (C-1, J_{C1.H1} 155.6), 76.4 (C-5), 71.5 (C-3), 70.4 (C-2), 65.1 (C-4), 62.4 (C-6), 30.9 (SAc), 20.8 $(4 \times \text{OAc})$; m/z (high-resolution ES-MS) 429.0830. C₁₆H₂₂O₁₀S· Na^+ requires *m/z*, 429.0826.

Methyl 1-thio-β-D-mannopyranoside 4

Compound 2 (100 mg, 246 µmol) was dissolved in anhydrous methanol (1.5 mL) and a solution of dry NaOMe-MeOH (700 µL; 3.5 M) was added under argon. The mixture was stirred for 30 min. Iodomethane (28 µL, 615 µmol) was then added dropwise to the mixture and the reaction was continued under argon for 1 h at room temperature. The mixture was neutralized with Dowex 50W (H⁺) resin and concentrated. Compound 4 was purified by reversed-phase chromatography using MeOHwater gradient (100%–20% MeOH) as eluent (47 mg, 91%); [a]_D²² -129.1 (c 1.1, MeOH) (Found: C, 38.3; H, 6.5. C₇H₁₄O₅S·0.5 H₂O requires C, 38.35; H, 6.4%); $\delta_{\rm H}$ (600 MHz; D₂O) 4.78 (d, 1H, J 0.9, H-1), 4.04 (dd, 1H, J 3.5, H-2), 3.92 (dd, 1H, J 2.4, 12.3, Ha-6), 3.73 (dd, 1H, J 6.2, Hb-6), 3.66 (dd, 1H, J 3.5, H-3), 3.60 (t, 1H, J 9.7, H-4), 3.41 (ddd, 1H, H-5), 2.27 (s, 3H, OMe); $\delta_{\rm C}$ (600 MHz; D₂O, from HMQC) 86.8 (C-1, $J_{\rm C1,H1}$ 156.4), 81.3 (C-5), 75.0 (C-3), 72.7 (C-2), 67.4 (C-4), 62.3 (C-6), 15.2 (OMe); m/z (HR ES-MS) 233.0456. C₇H₁₄O₂S·Na⁺ requires m/z, 233.0454.

Heptyl 1-thio-β-D-mannopyranoside 5

Under argon, a solution of anhydrous NaOMe-MeOH (840 μ L; 3.5 M) was added to compound 2 (100 mg, 246 μ mol) in MeOH (1.5 mL) and stirred for 30 min, 1-iodoheptane (162 µL, 984 µmol) was added dropwise, and the reaction was continued for 1 h at room temperature. After neutralization with Dowex 50W (H⁺) resin, the residue was purified by reversed-phase chromatography using a MeOH-water gradient (100-50% MeOH) to afford compound 5 (70 mg, 97%); [a]²²_D -73.0 (c 2.7, MeOH) (Found: C, 52.6; H, 9.1. C₁₃H₂₆O₅ requires C, 53.0; H, 8.9%; $\delta_{\rm H}$ (600 MHz; D₂O) 4.84 (s, 1H, H-1), 4.02 (d, 1 H, J 3.5, H-2), 3.91 (dd, 1H, J 2.2, 12.3, H^a-6), 3.73 (dd, 1H, J 6.2, H^b-6), 3.66 (dd, 1H, J 9.7, H-3), 3.60 (t, 1H, H-4), 3.40 (ddd, 1H, J 9.5, H-5), 2.75 (m, 2H, SCH₂), 1.64 (m, 2H, SCH₂CH₂), 1.38 [m, 2H, S(CH₂)₂CH₂], 1.35–1.24 [m, 6H, (CH₂)₃CH₃], 0.87 (t, 3H, J 6.8, CH₃); δ_C (600 MHz; D₂O, from HMQC) 85.0 (C-1, J_{C1.H1} 153.7), 81.0 (C-5), 74.8 (C-3), 73.2 (C-2), 67.5 (C-4), 62.0 (C-6), 31.7 (SCH₂), 30.0 (SCH₂CH₂), 28.5–13.3 [(CH₂)₄CH₃]; m/z (HR ES-MS) 317.1384. C₁₃H₂₆O₅S·Na⁺ requires *m*/*z*, 317.1393.

Methyl 6-(*p*-tolylsulfonyloxy)hexanoate 7

A solution of 5-(methoxycarbonyl)pentan-1-ol (6, 0.5 g, 3.4 mmol) in anhydrous pyridine (10 mL) was ice-cooled, toluene-

p-sulfonyl chloride (1.3 g, 6.8 mmol) was added, and the mixture was left at room temperature for 1 h. The mixture was then ice-cooled and water (1 mL) was added. The organic solvent was removed and the resulting residue was purified by chromatography on silica with hexane–EtOAc (4 : 1 v/v) to give compound 7 (0.77 g, 75%); $\delta_{\rm H}$ (300 MHz; CDCl₃) 7.76 (d, 2H, *J* 8.2, OTs), 7.32 (d, 2H, *J* 8.1 OTs), 4.00 (t, 2H, *J* 6.4, TsOCH₂), 3.63 (s, 3H, OMe), 2.43 (s, 3H, OTs), 2.22 (t, 2H, *J* 7.5, CH₂COOMe), 1.64 (m, 2H, TsOCH₂CH₂), 1.56 (m, 2H, CH₂CH₂COOMe), 1.33 [m, 2H, CH₂(CH₂)₂COOMe]; *m/z* (HR, ES-MS) 323.0934. C₁₄H₂₀O₅S·Na⁺ requires *m/z*, 323.0924.

5-(Methoxycarbonyl)pentyl 2,3,4,6-tetra-*O*-acetyl-1-thio-β-Dmannopyranoside 8 and 5-(methoxycarbonyl)pentyl 2,3,4,6-tetra-*O*-acetyl-1-thio-α-D-mannopyranoside 9

A solution of compound 2 (150 mg, 369 µmol) and tosylester 7 (140 mg, 503 µmol) in anhydrous DMF (2 mL) was cooled to -55 °C, diethylamine (200 µL) was added dropwise under argon, and the mixture was stirred for 48 h. The mixture was then diluted with EtOAc (30 mL) and the organic phase was washed successively with water $(1 \times 20 \text{ mL})$ and brine $(1 \times 20 \text{ mL})$ mL), dried over anhydrous Na2SO4, and concentrated. Chromatography on silica gel using hexane-AcOEt (7:3 v/v) as eluent gave, first, compound 9 (39 mg, 21%), then the β -isomer 8 (115 mg, 64%). Data for β -isomer 8: $[a]_D^{22}$ -55.4 (c 3.9, CHCl₃) (Found: C, 50.8; H, 6.7. C₂₁H₃₂O₁₁S requires C, 51.2, H, 6.55%); $\delta_{\rm H}$ (600 MHz; CDCl₃) 5.48 (d, 1H, J 3.5, H-2), 5.23 (t, 1H, J 10.1, H-4), 5.04 (dd, 1H, J 10.3, H-3), 4.72 (s, 1H, H-1), 4.24 (dd, 1H, J 5.9, 12.3, H^a-6), 4.12 (dd, 1H, J 2.4, H^b-6), 3.67 (ddd, 1H, H-5), 3.64 (s, 3H, OMe), 2.68 (t, 2H, J 7.5, SCH₂), 2.28 (t, 2H, J 7.5, CH₂COOMe), 2.16 (s, 3H, OAc), 2.06 (s, 3H, OAc), 2.02 (s, 3H, OAc), 1.95 (s, 3H, OAc), 1.61 (m, 4H, SCH₂CH₂-CH₂CH₂), 1.40 [m, 2H, CH₂(CH₂)₂COOMe]; $\delta_{\rm C}$ (600 MHz; CDCl₃, from HMQC) 82.7 (C-1, J_{C1,H1} 151.7), 76.8 (C-5), 72.0 (C-3), 70.4 (C-2), 66.0 (C-4), 62.9 (C-6), 51.8 (OMe), 34.0 (CH₂COOMe), 3.17 (SCH₂), 29.2 (CH₂CH₂COOMe), 28.1 [$CH_2(CH_2)_2COOMe$], 24.6 (SCH₂CH₂), 20.5 (4 × OAc); m/z (HR ES-MS) 515.1556. C₂₁H₃₂O₁₁S·Na⁺ requires m/z, 515.1558.

Data for α-isomer **9**: $[a]_{D}^{22}$ +44.7 (*c* 3.6, CHCl₃) (Found: C, 51.2; H, 6.5. C₂₁H₃₂O₁₁S requires C, 51.2; H, 6.55%); $\delta_{\rm H}$ (600 MHz; CDCl₃) 5.31 (dd, 1H, *J* 3.3, H-2), 5.29 (t, 1H, *J* 9.9, H-4), 5,24 (dd, 1H, *J* 10.1, H-3), 5.23 (d, 1H, *J* 1.7, H-1), 4.35 (ddd, 1H, H-5), 4.29 (dd, 1H, *J* 5.3, 12.3, H^a-6), 4.07 (dd, 1H, *J* 2.4, 12.3, H^b-6), 3.65 (s, 3H, OMe), 2.60 (m, 2H, SCH₂), 2.29 (t, 2H, *J* 7.3, CH₂COOMe), 2.14 (s, 3H, OAc), 2.07 (s, 3H, OAc), 2.02 (s, 3H, OAc), 1.97 (s, 3H, OAc), 1.62 (m, 4H, SCH₂CH₂-CH₂CH₂), 1.39 [m, 2H, S(CH₂)₂CH₂]; *m*/*z* (HR ES-MS) 515.1556. C₂₁H₃₂O₁₁S·Na⁺ requires *m*/*z*, 515.1558.

5-(Methoxycarbonyl)pentyl 1-thio-β-D-mannopyranoside 10

Compound 8 (100 mg, 203 µmol) was deprotected by transesterification in anhydrous methanol (20 mL) containing a catalytic amount of NaOMe, and the product 10 was purified by reversed-phase chromatography using MeOH-water gradient (100–50% MeOH) as eluent (60 mg, 91%); $[a]_{D}^{22}$ –73.1 (c 1.3, MeOH) (Found: C, 48.5; H, 7.0. C₁₃H₂₄O₇S requires C, 48.1; H, 7.5%); $\delta_{\rm H}$ (600 MHz; D₂O) 4.74 (d, 1H, J 0.9, H-1), 3.96 (dd, 1H, J 3.5, H-2), 3.84 (dd, 1H, J 2.2, 12.3, H^a-6), 3.66 (dd, 1H, J 6.2, H^b-6), 3.63 (s, 3H, OMe), 3.60 (dd, 1H, J 9.7, H-3), 3.53 (t, 1H, J 9.9, H-4), 3.34 (ddd, 1H, H-5), 2.69 (m, 2H, SCH₂), 2.34 (t, 2H, J 7.3, CH₂COOMe), 1.58 (m, 4H, SCH₂CH₂CH₂CH₂), 1.35 [m, 2H, S(CH₂)₂CH₂]; $\delta_{\rm C}$ (600 MHz; CDCl₃, from HMQC) 85.3 (C-1, J_{C1,H1} 154.0), 79.0 (C-5), 74.8 (C-3), 73.2 (C-2), 67.5 (C-4), 62.0 (C-6), 53.0 (OMe), 34.7 (CH₂COOMe), 31.2 (SCH₂), 29.6 (CH₂CH₂COOMe), 28.2 [S(CH₂)₂CH₂], 24.6 $(SCH_2CH_2); m/z$ (HR ES-MS) 347.1143. $C_{13}H_{24}O_7S\cdot Na^+$ requires m/z, 347.1135.

Methyl 2,3-di-O-acetyl-4-O-benzoyl-6-S-(2,3,4,6-tetra-O-acetyl-β-D-mannopyranosyl)-6-thio-α-D-glucopyranoside 12

A mixture of compound 2 (60 mg, 148 µmol) and methyl 2,3-di-O-acetyl-6-bromo-4-O-benzoyl-6-deoxy-α-D-glucopyranside 11 (100 mg, 225 µmol) in anhydrous DMF (2 mL) was cooled to -15 °C under argon, diethylamine (300 µL) was added dropwise, and the mixture was stirred for 24 h. The reaction mixture was diluted with EtOAc (30 mL) and the organic phase was washed with water $(2 \times 15 \text{ mL})$, dried over anhydrous Na₂SO₄, and concentrated. Compound 12 was obtained by chromatography on silica gel using toluene–AcOEt (7: 3 v/v) as eluent (80 mg, 74%). This compound was contaminated with a trace amount of the α -isomer ($\approx 6\%$, as judged from NMR); $[a]_D^{22}$ +18.4 (c 5.7, CHCl₃) (Found: C, 52.3; H, 5.7; S, 4.6. C₃₂H₄₀O₁₇S requires C, 52.7; H, 5.5; S, 4.4%); $\delta_{\rm H}$ (600 MHz; CDCl₃) 7.94-7.42 (m, 5H, Bz), 5.63 (t, 1H, J 9.7, H-3), 5.50 (d, 1H, J 3.5, H-2'), 5.19 (t, 1H, J 10.1, H-4'), 5.12 (t, 1H, J 9.7, H-4), 5.00 (dd, 1H, J 10.1, H-3'), 4.98 (d, 1H, J 3.7, H-1), 4.92 (dd, 1H, J 10.1, H-2), 4.91 (s, 1H, H-1'), 4.19 (dd, 1H, J 6.2, 12.5, H^a-6'), 4.08–4.05 (m, 2H, H^b-6', H-5), 3.63 (ddd, 1H, H-5'), 3.46 (s, 3H, OMe), 2.88 (dd, 1H, J 9.3, 14.3, Ha-6), 2.83 (dd, 1H, J 2.8, H^b-6), 2.13 (s, 3H, OAc), 2.07 (s, 3H, OAc), 2.02 (s, 6H, $2 \times OAc$), 1.94 (s, 3H, OAc), 1.86 (s, 3H, OAc); $\delta_{\rm C}$ (600 MHz; CDCl₃, from HMQC) 133.8–128.2 (Bz), 96.5 (C-1, J_{C1,H1} 171.9), 83.6 (C-1', J_{C1,H1'} 154.0), 76.6 (C-5'), 72.4 (C-4), 71.8 (C-3'), 70.8 (C-2), 70.2 (C-2', C-5), 69.4 (C-3), 65.5 (C-4'), 62.8 (C-6'), 56.0 (Me), 33.9 (C-6), 20.7 (8 \times OAc); m/z (HR ES-MS) 751.1874. C₃₂H₄₀O₁₇S·Na⁺ requires m/z, 751.1884.

Methyl 6-S-(β-D-mannopyranosyl)-6-thio-α-D-glucopyranoside 13

Compound **12** (54 mg, 74.1 µmol) was transesterified and purified as described for **8** to yield *compound* **13** in pure form (25 mg, 92%); $[a]_{D}^{22} + 23.3$ (*c* 3.0, MeOH) (Found: C, 40.2; H, 6.5. $C_{13}H_{24}O_{10}S \cdot H_2O$ requires C, 40.0; H, 6.7%); δ_H (600 MHz; D₂O) 4.95 (d, 1H, *J* 0.7, H-1'), 4.79 (d, 1H, *J* 3.8, H-1), 4.07 (dd, 1H, *J* 3.5, H-2'), 3.92 (dd, 1H, *J* 12.3, H^a-6'), 3.82 (m, 1H, H-5), 3.75 (dd, 1H, *J* 6.1, H^b-6'), 3.66 (dd, 1H, *J* 3.5, 9.5, H-3'), 3.64 (t, 1H, *J* 8.6, H-4'), 3.61 (t, 1H, *J* 9.7, H-3), 3.58 (dd, 1H, *J* 9.7, H-2), 3.45 (s, 3H, OMe), 3.41 (ddd, 1H, H-5'), 3.33 (t, 1H, *J* 9.2, H-4), 3.25 (dd, 1H, *J* 14.1, H^a-6), 2.93 (dd, 1H, *J* 8.6, H^b-6); δ_C (600 MHz; CDCl₃, from HMQC) 100.1 (C-1, $J_{C1,H1}$ 168.2), 86.5 (C-1', $J_{C'1,H'1}$ 154.1), 81.4 (C-5'), 74.7 (C-3'), 74.0 (C-4'), 73.7 (C-2'), 72.2 (C-5, C-2), 67.5 (C-3), 62.0 (C-6'), 56.0 (Me), 33.9 (C-6); *m/z* (HR ES-MS) 395.0984. $C_{13}H_{24}O_{10}S \cdot Na^+$ requires *m/z*, 395.0982.

Methyl 2,3,6-tri-*O*-benzoyl-4-*S*-(2,3,4,6-tetra-*O*-acetylβ-D-mannopyranosyl)-4-thio-β-D-glucopyranoside 15

Under argon, diethylamine (500 µL) was added dropwise to a stirred solution of triflate 14 (202 mg, 316 µmol) and thioacetate 2 (117 mg, 288 µmol) in anhydrous DMF (4 mL) at -5 °C, and reaction was continued for 12 h. EtOAc (100 mL) was added and the resulting solution was washed successively with water $(1 \times 30 \text{ mL})$ and brine $(1 \times 30 \text{ mL})$, dried over anhydrous Na₂SO₄, and concentrated. The residue was purified by chromatography on silica using toluene-EtOAc (7:3 v/v) as eluent to give compound 15 (196 mg, 80%). This compound was contaminated with the α -isomer ($\approx 5\%$, as judged from NMR); [a]_D²² - 3.4 (c 8.3, CHCl₃) (Found: C, 58.9; H, 5.3. $C_{42}H_{44}O_{17}S$ requires C, 59.15; H, 5.2%); δ_{H} (600 MHz; C₆D₆) 8.20-6.86 (m, 15H, Bz), 5.99-5.92 (m, 2H, H-3, H-2), 5.82 (dd, 1H, J 3.5, H-2'), 5.50 (t, 1H, J 10.1, H-4'), 5.42 (dd, 1H, J 10.1, H-3'), 5.14 (d, 1H, J 1.1, H-1'), 5.09 (dd, 1H, J 2.4, 12.1, H^a-6), 4.95 (dd, 1H, J 5.3, H^b-6), 4.67 (d, 1H, J 7.5, H-1), 4.16 (dd, 1H, J 2.4, 12.3, H^a-6'), 4.07 (dd, 1H, J 6.9, H^b-6'), 4.00 (ddd, 1H, H-5), 3.26 (s, 3H, OMe), 3.21 (ddd, 1H, H-5'), 3.10 (t, 1H, J 10.6, H-4), 1.75 (s, 3H, OAc), 1.69 (s, 3H, OAc), 1.64 (s, 3H, OAc), 1.56 (s, 3H, OAc); $\delta_{\rm C}$ (600 MHz; CDCl₃, from HMQC) 132.5–129.5 (Bz), 102.0 (C-1, $J_{\rm C1,H1}$ 157.7), 80.5 (C-1', $J_{\rm C1,H1}$, 150.5), 76.5 (C-5'), 74.3 (C-5), 73.3 (C-2), 71.6 (C-3'), 71.2 (C-3), 70.2 (C-2'), 65.8 (C-4'), 64.0 (C-6), 62.0 (C-6'), 55.3 (OMe), 46.6 (C-4), 20.0 (4 × OAc); m/z (HR ES-MS) 875.2199. C₄₂H₄₄O₁₇S·Na⁺ requires m/z, 875.2197.

Methyl 4-S-(β-D-mannopyranosyl)-4-thio-β-D-glucopyranoside 16

Compound **15** (50 mg, 58.6 µmol) was deprotected and purified as described for **8** to yield *compound* **16** in pure form (20 mg, 93%); $[a]_{D}^{22}$ -40.2 (*c* 2.5, MeOH) (Found: C, 37.3; H, 6.9. C₁₃H₂₄O₁₀S·2.5H₂O requires C, 37.4; H, 7.0%); δ_{H} (600 MHz; D₂O) 4.95 (s, 1H, H-1'), 4.36 (d, 1H, *J* 8.1, H-1), 4.15 (dd, 1H, *J* 2.0, 12.3, H^a-6), 4.07 (d, 1H, *J* 3.5, H-2'), 3.90 (dd, 1H, *J* 5.1, H^b-6), 3.89 (dd, 1H, *J* 2.4, 12.4, H^a-6'), 3.73–3.68 (m, 2H, H^b-6, H-5), 3.67 (dd, 1H, *J* 9.7, H-3'), 3.63 (dd, 1H, *J* 10.4, H-3), 3.59 (t, 1H, *J* 9.9, H-4'), 3.57 (s, 3H, OMe), 3.42 (ddd, 1H, *J* 2.2, 6.4, 9.3, H-5'), 3.3 (t, 1H, *J* 8.6, H-2), 2.82 (t, 1H, *J* 10.8, H-4); δ_{C} (600 MHz; D₂O, from HMQC) 104.0 (C-1, *J*_{C1,H1} 160.8), 86.0 (C-1', *J*_{C1,H1}, 153.5), 81.1 (C-5'), 77.2 (C-5), 75.3 (C-2), 74.5 (C-3'), 73.5 (C-2'), 67.5 (C-4'), 62.0 (C-6, C-6'), 57.9 (OMe), 48.5 (C-4); *m/z* (HR ES-MS) 395.0980. C₁₃H₂₄O₁₀-S·Na⁺ requires *m/z*, 395.0982.

Heptyl 2,4-di-*O*-benzoyl-1-thio-β-D-mannopyranoside 19, heptyl 2,6-di-*O*-benzoyl-1-thio-β-D-mannopyranoside 20 and heptyl 2-*O*-benzoyl-1-thio-β-D-mannopyranoside 21

CSA (40 mg, 0.17 mmol) was added to a stirred suspension of 5 (300 mg, 1.02 mmol) and trimethyl orthobenzoate (700 μ L, 4.1 mmol) in CHCl₃ (60 mL). After the mixture became clear, the solution was concentrated on a rotovapor to ≈ 10 mL, CHCl₃ (50 mL) was added, and the solution was concentrated again to ≈ 10 mL volume. This process was repeated until TLC showed that all the starting material was consumed. Triethylamine (3 mL) was added to the reaction mixture and the solution was evaporated to dryness. A solution of 90% acetic acid–water (10 mL) was added to the reaction flask and reaction was continued for 30 min. After evaporation, the mixture was purified by column chromatography (toluene – EtOAc 4 : 1 \rightarrow 1 : 1 v/v) to give, first, **19** (166 mg, 32%), then **20** (204 mg, 40%), and last **21** (51 mg, 13%).

Data for *dibenzoate* **19**: $[a]_{D}^{22}$ –94.0 (*c* 5.7, CHCl₃) (Found: C, 64.4; H, 6.7. $C_{27}H_{34}O_7S$ requires C, 64.5; H, 6.8%); δ_H (600 MHz; CDCl₃) 8.06–7.44 (m, 10H, Bz), 5.75 (dd, 1H, *J* 3.6, H-2), 5.38 (t, 1H, *J* 9.5, H-4), 4.88 (d, 1H, *J* 0.7, H-1), 4.18 (dd, 1H, *J* 9.8, H-3), 3.86–3.68 (m, 3H, H-5, H-6), 2.70 (t, 2H, *J* 7.5, SCH₂), 1.60 (m, 2H, SCH₂CH₂), 1.38–1.20 [m, 8H, (CH₂)₄CH₃], 0.85 (t, 3H, *J* 6.8, CH₃); δ_C (600 MHz; CDCl₃, from HMQC) 133.8–128.6 (Bz), 83.3 (C-1, *J*_{C1,H1} 149.2), 78.8 (C-5), 73.9 (C-2), 73.0 (C-3), 70.4 (C-4), 62.1 (C-6), 31.9 (SCH₂), 31.7 (SCH₂CH₂), 29.8 [S(CH₂)₂CH₂], 28.9 [S(CH₂)₃-CH₂], 22.6 (CH₂CH₂CH₃), 21.0 (CH₂CH₃), 14.1 (CH₃); *m/z* (HR ES-MS) 525.1932. $C_{27}H_{34}O_7S\cdotNa^+$ requires *m/z*, 525.1932.

Data for *dibenzoate* **20**: $[a]_{D}^{22} - 70.9$ (*c* 3.4, CHCl₃) (Found: C, 64.4; H, 6.7. $C_{27}H_{34}O_7S$ requires C, 64.5; H, 6.8%); δ_H (600 MHz, CDCl₃) 8.11–7.24 (m, 10H, Bz), 5.63 (d, 1H, J 3.5, H-2), 4.83 (s, 1H, H-1), 4.79 (dd, 1H, J 4.8, 12.1, H^a-6), 4.62 (dd, 1H, J 2.0, H^b-6), 3.93 (dd, 1H, J 9.3, H-3), 3.82 (t, 1H, J 9.5, H-4), 3.67 (ddd, 1H, J 2.2, 4.8, 9.5, H-5), 2.69 (m, 2H, SCH₂), 1.61–1.20 [m, 10H, SCH₂(CH₂)₅], 0.83 (t, 3H, J 7.0, CH₃); δ_C (600 MHz; CDCl₃, from HMQC) 130.0–128.2 (Bz), 83.0 (C-1, $J_{C1,H1}$ 153.7), 79.0 (C-5), 75.0 (C-3), 72.8 (C-2), 68.0 (C-4), 64.4 (C-6), 31.5 (SCH₂), 28.9–14.2 [SCH₂(CH₂)₅]; *m/z* (HR ES-MS) 635.1070. $C_{27}H_{34}O_7S \cdot Cs^+$ requires *m/z*, 635.1074. Data for *monobenzoate* **21**: $[a]_{D}^{22} - 51.2$ (*c* 4.3, CHCl₃) (Found: C, 60.1; H, 7.7; S, 7.9. $C_{20}H_{30}O_6S$ requires C, 60.3; H, 7.6; S, 8.05%); δ_H (600 MHz; CDCl₃) 8.08–7.42 (m, 5H, Bz), 5.60 (s, 1H, H-2), 4.83 (s, 1H, H-1), 3.97 (dd, 1H, *J* 3.7, 12.1, H^a-6), 3.89–3.87 (m, 3H, H-3, H-4, H^b-6), 3.44 (m, 1H, H-5), 2.69 (t, 2H, *J* 7.5, SCH₂), 1.93 (br, 3H, 3 × OH), 1.58 (m, 2H, SCH₂CH₂), 1.35–1.22 [m, 8H, (CH₂)₄CH₃], 0.85 (t, 3H, *J* 7.0, CH₃); δ_C (600 MHz; CDCl₃, from HMQC) 167.1, 133.6, 130.2, 129.2, 128.6 (Bz), 83.2 (C-1), 79.8 (C-5), 77.5 (C-2), 74.1 (C-3), 68.6 (C-4), 62.8 (C-6), 31.9 (SCH₂), 31.7 (SCH₂CH₂), 29.9 [S(CH₂)₂CH₂], 28.9 [S(CH₂)₃CH₂], 28.8 (CH₂CH₂CH₃), 22.6 (CH₂CH₃), 14.1 (CH₃); *m/z* (HR ES-MS) 421.1658. $C_{20}H_{30}$ - O_6S -Na⁺ requires *m/z*, 421.1655.

Heptyl 2,4-di-*O*-benzoyl-3,6-bis-*O*-(2,3,4,6-tetra-*O*-acetyl-α-Dmannopyranosyl)-1-thio-β-D-mannopyranoside 23 and heptyl 2,4-di-*O*-benzoyl-6-*O*-(2,3,4,6-tetra-*O*-acetyl-α-D-mannopyranosyl)-1-thio-β-D-mannopyranoside 24

A solution of compound **19** (100 mg, 199 µmol) and 2,3,4,6tetra-*O*-acetyl- α -D-mannopyranosyl trichloroacetimidate **22** (262 mg, 531 µmol) in anhydrous CH₂Cl₂ (6 mL) was cooled to 0 °C under argon. Boron trifluoride–diethyl ether (40 µL, 316 µmol) was added dropwise to the solution and the reaction mixture was allowed to warm slowly and stirred for 5 h at room temperature. The reaction was quenched with triethylamine (2 mL) and the solvent was evaporated. The mixture was purified by chromatography on silica gel using hexane–EtOAc (7 : 3 v/v) as eluent to afford, first, disaccharide **24** (80 mg, 48%) and then the *trisaccharide* **23** (119 mg, 51%).

Data for compound 23: [a]_D²² -30.5 (c 2.1, CHCl₃) (Found: C, 56.6; H, 5.9. C₅₅H₇₀O₂₅S requires C, 56.8; H, 6.1%); $\delta_{\rm H}$ (600 MHz; CDCl₃) 8.12–7.41 (m, 10H, Bz), 5.78 (dd, 1H, J 3.5, H-2), 5.51 (t, 1H, J 9.9, H-4), 5.28 (dd, 1H, J 10.1, H-3"), 5.24 (t, 1H, J 9.9, H-4"), 5.17 (dd, 1H, J 1.8, 3.3, H-2"), 5.12 (t, 1H, J 9.7, H-4'), 5.01 (dd, 1H, J 9.7, H-3'), 4.92 (d, 1H, J 2.6, H-1'), 4.89 (d, 1H, J 0.9, H-1), 4.83 (dd, 1H, J 3.5, H-2'), 4.77 (d, 1H, J 1.7, H-1"), 4.30 (dd, 1H, J 2.2, 4.8, H-5'), 4.26 (dd, 1H, J 12.3, H^a-6'), 4.23 (dd, 1H, J 10.2, H-3), 4.21 (dd, 1H, J 4.8, 11.9, Ha-6"), 4.17 (dd, 1H, J 2.0, Hb-6), 4.07 (ddd, 1H, H-5"), 4.03 (dd, 1H, J 2.4, H^b-6"), 3.95 (dd, 1H, J 7.1, 10.8, H^a-6), 3.85 (ddd, 1H, J 2.4, H-5), 3.60 (dd, 1H, H^b-6), 2.72 (m, 2H, SCH₂), 2.11 (s, 3H, OAc), 2.10 (s, 3H, OAc), 2.02 (s, 3H, OAc), 2.00 (s, 3H, OAc), 1.99 (s, 3H, OAc), 1.95 (s, 3H, OAc), 1.79 (s, 3H, OAc), 1.78 (s, 3H, OAc), 1.60 [m, 4H, SCH₂(CH₂)₂], 1.36–1.19 [m, 6H, $(CH_2)_3CH_3$], 0.85 (t, 3H, J 7.1, CH₃); δ_C (600 MHz; CDCl₃, from HMQC) 132.3–128.5 (Bz), 99.4 (C-1', J_{C1,H1}, 172.3), 97.0 (C-1", J_{C1",H1"} 172.6), 82.7 (C-1, J_{C1,H1} 150.5), 77.2 (C-5, C-6"), 72.4 (C-2), 69.3 (C-4, C-5'), 69.2 (C-2"), 69.0 (C-2'), 68.9 (C-4"), 68.2 (C-5"), 68.0 (C-3'), 67.4 (C-6), 65.7 (C-3", C-4'), 62.4 (C-3, C-6'), 31.4–22.4 [S(CH₂)₆], 20.6–20.2 $(8 \times \text{OAc})$, 13.8 (CH₃); m/z (HR ES-MS) 1185.3810. C₅₅H₇₀-O₂₅S·Na⁺ requires *m*/*z*, 1185.3819.

Data for compound **24**: $[a]_{22}^{22} - 12.1$ (c 6.6, CHCl₃) (Found: C, 59.0; H, 3.6. C₄₁H₅₂O₁₆S requires C, 59.1; H, 6.3%); $\delta_{\rm H}$ (600 MHz; CDCl₃) 8.14–7.42 (m, 10H, Bz), 5.75 (d, 1H, J 3.5, H-2), 5.38 (t, 1H, J9.7, H-4), 5.31–5.23 (m, 3H, H-2', H-3', H-4'), 4.88 (s, 1H, H-1), 4.81 (d, 1H, J 1.5, H-1'), 4.15 (dd, 1H, J 9.7, H-3), 4.13 (dd, 1H, J 4.8, 12.1, H^a-6'), 3.99–3.89 (m, 4H, H-5', H^a-6, H^b-6 + H-5), 3.66 (dd, 1H, J 1.8 H, 10.6, H^b-6), 2.72 (m, 2H, SCH₂), 2.11 (s, 3H, OAc), 2.00 (s, 3H, OAc), 1.99 (s, 3H, OAc), 1.96 (s, 3H, OAc), 1.64–1.20 [m, 10H, (CH₂)₅CH₃], 0.85 (t, 3H, J 7.1, CH₃); $\delta_{\rm C}$ (600 MHz; CDCl₃, from HMQC) 133.6–128.7 (Bz), 97.1 (C-1', J_{C1,H1}, 170.2), 82.8 (C-1, J_{C1,H1} 149.5), 76.8 (C-5), 73.7 (C-2), 73.0 (C-3), 70.5 C-4, 69.0 (C-2', C-3'), 68.3 (C-5'), 67.0 (C-6), 65.8 (C-4'), 62.0 (C-6'), 31.4–22.4 [S(CH₂)₆], 20.8–20.4 (4 × OAc); 13.8 (CH₃); m/z (HR ES-MS) 855.2873. C₄₁H₅₂O₁₆S·Na⁺ requires m/z, 855.2868.

Heptyl 3,6-di-*O*-(α-D-mannopyranosyl)-1-thio-β-D-mannopyranoside 25

Compound 23 (110 mg, 95 µmol) was transesterified and the crude product was purified as described for 8 to yield compound 25 (53 mg, 91%); [a]²²_D +18.4 (c 2.6, MeOH) (Found: C, 45.55; H, 7.5. C₂₅H₄₆O₁₅S·2H₂O requires C, 45.85; H, 7.7%); $\delta_{\rm H}$ (600 MHz; D₂O) 5.11 (d, 1H, J 1.6, H-1'), 4.90 (d, 1H, J 1.7, H-1"), 4.85 (s, 1H, H-1), 4.19 (d, 1H, J 3.5, H-2), 4.07 (dd, 1H, J 3.5, H-2'), 3.99 (dd, 1H, J 3.5, H-2"), 3.95 (dd, 1H, J 5.1, 11.4, Ha-6), 3.92-3.87 (m, 3H, H-3', Ha-6', Ha6a"), 3.83 (dd, 1H, J 9.0, H-3"), 3.82 (t, 1H, J 9.7, H-4), 3.79-3.73 (m, 5H, H-3, H-4', Hb-6, Hb-6', Hb-6"), 3.72-3.65 (m, 3H, H-5", H-5', H-4"), 3.60 (ddd, 1H, J 1.8, H-5), 2.74 (m, 2H, SCH₂), 1.64 (m, 2H, SCH₂CH₂), 1.42–1.24 [m, 8H, (CH₂)₄], 0.87 (t, 3H, J 7.0, CH₃); $\delta_{\rm C}$ (600 MHz; D₂O, from HMQC) 103.2 (C-1', J_{C1,H1}, 172.4), 100.2 (C-1", J_{C1",H1"} 171.6), 85.9 (C-1, J_{C1,H1} 154.1), 82.7 (C-4'), 79.2 (C-5), 74.2 (C-3), 73.4 (C-5"), 72.8 (C-2), 71.6 (C-4), 71.2 (C-3'), 71.0 (C-2'), 70.9 (C-2"), 67.6 (C-4", C-5'), 66.6 (C-3"), 66.4 (C-6), 61.9 (C-6', C-6"), 31.8 (SCH₂), 30.2 (SCH₂CH₂), 32.2–23.0 [(CH₂)₄], 14.4 (CH₃); *m*/*z* (HR ES-MS) 641.2446. $C_{25}H_{46}O_{15}S \cdot Na^+$ requires *m*/*z*, 641.2450.

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